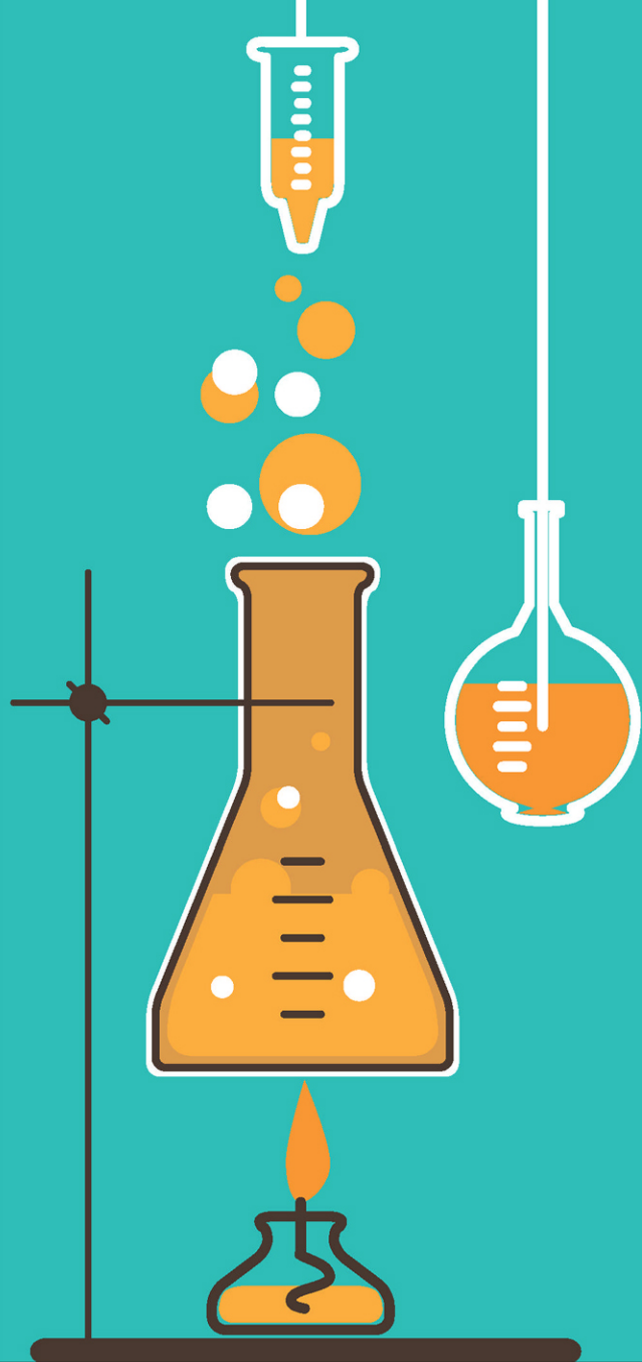


GLOBAL  
EDITION



# Introductory Chemistry Essentials

SIXTH EDITION in SI Units

Nivaldo J. Tro



# Periodic Table of the Elements

GROUP											
1											
1A											
PERIOD	1	1 <b>H</b> 1.01 hydrogen	2 2A								
	2	3 <b>Li</b> 6.94 lithium	4 <b>Be</b> 9.01 beryllium								
	3	11 <b>Na</b> 22.99 sodium	12 <b>Mg</b> 24.31 magnesium	3 3B	4 4B	5 5B	6 6B	7 7B	8 8B	9 8B	
	4	19 <b>K</b> 39.10 potassium	20 <b>Ca</b> 40.08 calcium	21 <b>Sc</b> 44.96 scandium	22 <b>Ti</b> 47.88 titanium	23 <b>V</b> 50.94 vanadium	24 <b>Cr</b> 52.00 chromium	25 <b>Mn</b> 54.94 manganese	26 <b>Fe</b> 55.85 iron	27 <b>Co</b> 58.93 cobalt	
	5	37 <b>Rb</b> 85.47 rubidium	38 <b>Sr</b> 87.62 strontium	39 <b>Y</b> 88.91 yttrium	40 <b>Zr</b> 91.22 zirconium	41 <b>Nb</b> 92.91 niobium	42 <b>Mo</b> 95.95 molybdenum	43 <b>Tc</b> (99) technetium	44 <b>Ru</b> 101.07 ruthenium	45 <b>Rh</b> 102.91 rhodium	
	6	55 <b>Cs</b> 132.91 cesium	56 <b>Ba</b> 137.33 barium	57 <b>La</b> 138.91 lanthanum	72 <b>Hf</b> 178.49 hafnium	73 <b>Ta</b> 180.95 tantalum	74 <b>W</b> 183.85 tungsten	75 <b>Re</b> 186.21 rhenium	76 <b>Os</b> 190.23 osmium	77 <b>Ir</b> 192.22 iridium	
	7	87 <b>Fr</b> (223) francium	88 <b>Ra</b> (226) radium	89 <b>Ac</b> (227) actinium	104 <b>Rf</b> (261) rutherfordium	105 <b>Db</b> (262) dubnium	106 <b>Sg</b> (263) seaborgium	107 <b>Bh</b> (262) bohrium	108 <b>Hs</b> (265) hassium	109 <b>Mt</b> (266) meitnerium	
Lanthanide series				58 <b>Ce</b> 140.12 cerium	59 <b>Pr</b> 140.91 praseodymium	60 <b>Nd</b> 144.24 neodymium	61 <b>Pm</b> (147) promethium	62 <b>Sm</b> 150.36 samarium	63 <b>Eu</b> 151.97 europium		
Actinide series				90 <b>Th</b> (232) thorium	91 <b>Pa</b> (231) protactinium	92 <b>U</b> (238) uranium	93 <b>Np</b> (237) neptunium	94 <b>Pu</b> (244) plutonium	95 <b>Am</b> (243) americium		

\*The mass number of an important radioactive isotope—not the atomic mass—is shown in parentheses for those elements with no stable isotopes.

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